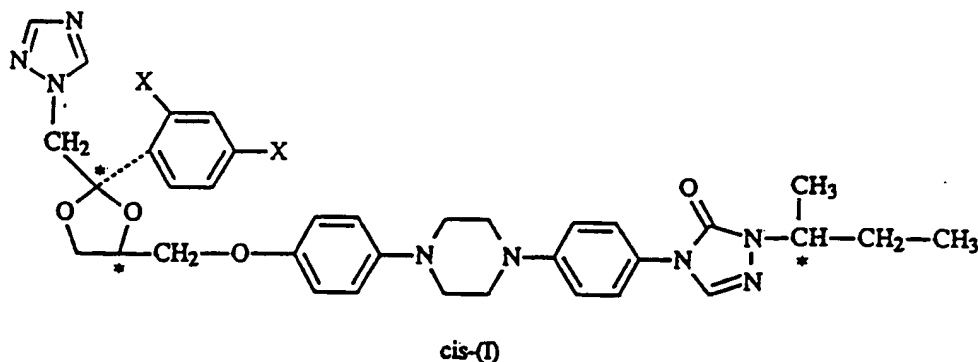




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(54) Title: ITRACONAZOLE AND SAPERCONAZOLE STEREOISOMERS

**(57) Abstract**

The present invention is concerned with the stereoisomeric forms of itraconazole (X=Cl) and saperconazole (X=F), which may be represented by formula (cis-(I)), and the pharmaceutically acceptable acid addition salt forms thereof, processes for preparing said stereoisomeric forms, the complexes thereof with cyclodextrin derivatives, pharmaceutical compositions comprising said complexes and methods of preparing said complexes and pharmaceutical compositions.

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ITRACONAZOLE AND SAPERCONAZOLE STEREOISOMERS

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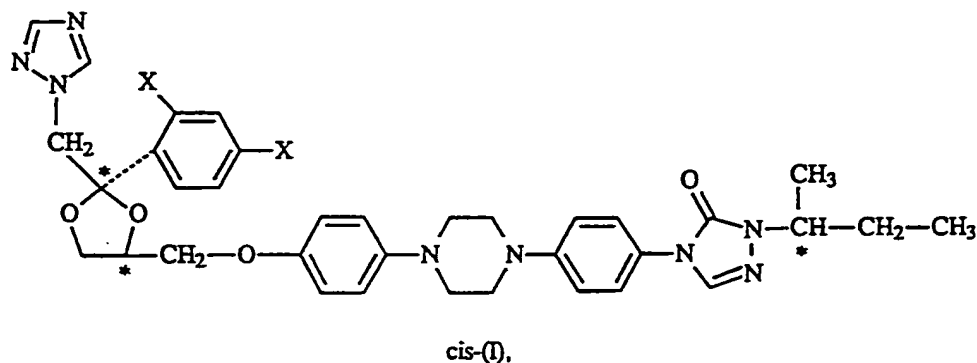
The present invention is concerned with the stereoisomeric forms of itraconazole and saperconazole, processes for preparing said stereoisomeric forms, the complexes thereof with cyclodextrin derivatives, pharmaceutical compositions comprising said complexes and methods of preparing said complexes and pharmaceutical compositions.

Itraconazole or (+)-cis-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methyl-propyl)-3H-1,2,4-triazol-3-one, is a broadspectrum antifungal compound developed for oral, parenteral and topical use and is disclosed in US-4,267,179. Its difluoro analog, saperconazole or (+)-cis-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-yl-methyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]-phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one, has improved activity against *Aspergillus* spp. and is disclosed in US-4,916,134. Both compounds exist as a mixture of four stereoisomers.

The development of efficacious pharmaceutical compositions of itraconazole and saperconazole is hampered considerably by the fact that said compounds are only very sparingly soluble in water. The solubility of both compounds can be increased by complexation with cyclodextrins or derivatives thereof as described in WO 85/02767 and US-4,764,604.

Unexpectedly, it has now been found that each of the individual stereoisomers of itraconazole and saperconazole have greater water solubility than the diastereomeric mixtures of said compounds, in particular when complexed with cyclodextrin or its derivatives. As a result, pharmaceutical compositions having good bioavailability, yet comprising less cyclodextrin as a complexing agent, can be prepared.

The present invention is concerned with the stereoisomeric forms of itraconazole (X = Cl) and saperconazole (X = F), which may be represented by the formula



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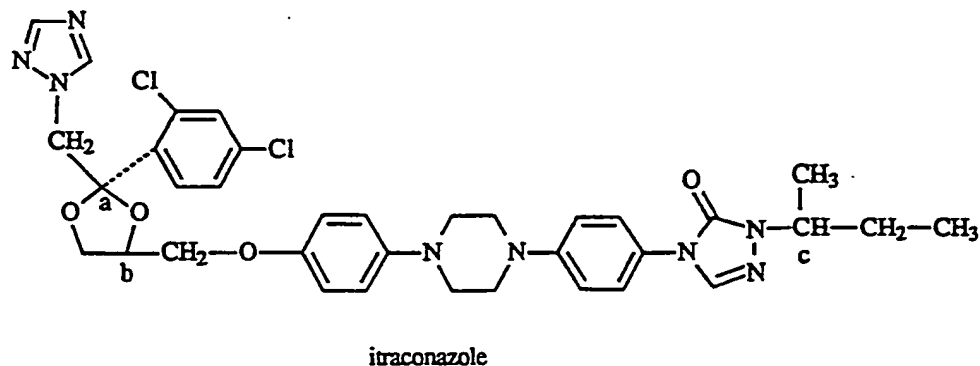
and the pharmaceutically acceptable acid addition salt forms thereof. The three asterisks indicate the three chiral centers, and 'cis' means that the (1H-1,2,4-triazol-1-ylmethyl) moiety and the substituted phenoxy moiety are located at the same side of the plane defined by the 1,3-dioxolane ring.

10

The four possible stereoisomeric cis forms can be described using various rules of nomenclature. The following tables show the correlation among the C.A. stereochemical descriptor, the absolute configuration at each of the chiral centers and the specific optical rotation $[\alpha]_D^{20}$ in 1% methanol (itraconazole; table I) (saperconazole; table II).

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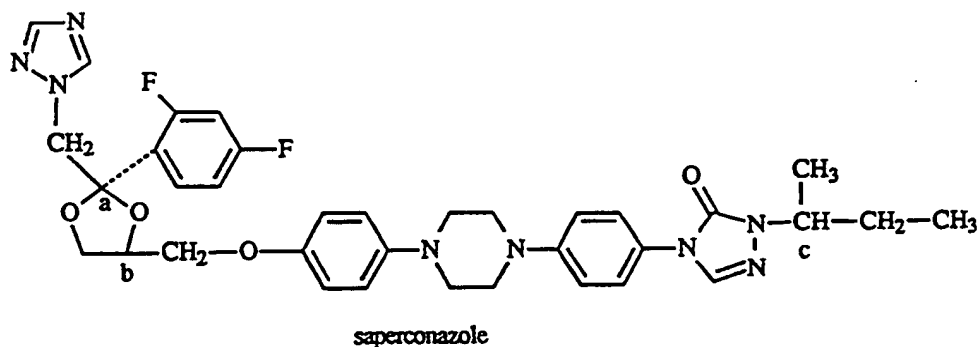
Table I



C.A. descriptor	absolute configuration at			$[\alpha]_D^{20}$ (1% CH ₃ OH)
	(a)	(b)	(c)	
(+)-[2R-[2 α ,4 α ,4(R)]]	R	S	R	+14.15°
(+)-[2R-[2 α ,4 α ,4(S)]]	R	S	S	+19.08°
(-)-[2S-[2 α ,4 α ,4(R)]]	S	R	R	-18.78°
(-)-[2S-[2 α ,4 α ,4(S)]]	S	R	S	-13.46°

Table II

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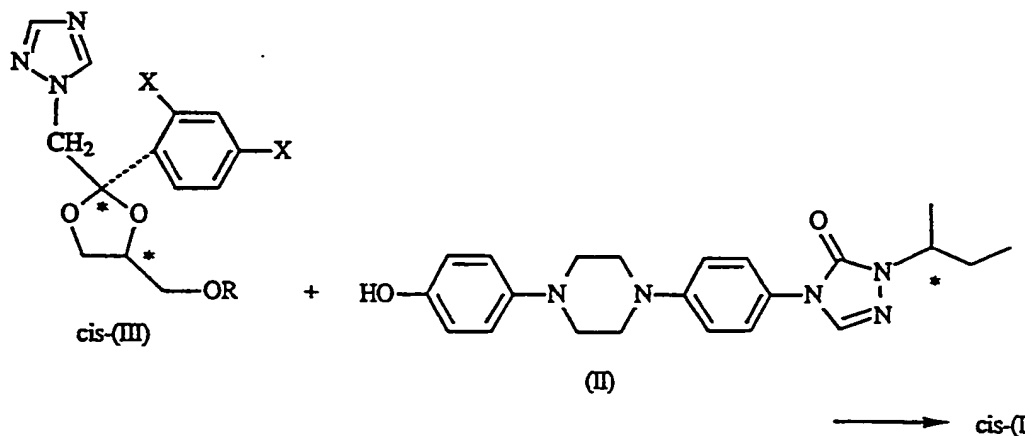
C.A. descriptor	absolute configuration at			$[\alpha]_D^{20}$ (1% CH ₃ OH)
	(a)	(b)	(c)	
(+)-[2R-[2 α ,4 α ,4(R)]]	R	S	R	+9.00°
(+)-[2R-[2 α ,4 α ,4(S)]]	R	S	S	+14.13°
(-)-[2S-[2 α ,4 α ,4(R)]]	S	R	R	-13.55°
(-)-[2S-[2 α ,4 α ,4(S)]]	S	R	S	-8.41°

- 10 The term 'stereoisomeric form' as used herein concerns compounds having a stereoisomeric purity of at least 96% up to a stereoisomeric purity of 100%, in particular compounds having a stereoisomeric purity of 98% up to 100%. In particular, said stereoisomeric forms define compounds having an enantiomeric excess and a diastereomeric excess of at least 96% up to 100%, more particularly, having an
- 15 enantiomeric excess and a diastereomeric excess of 98% up to 100%.

With regard to the intermediates described hereinafter, the term "enantiomerically pure" defines intermediates having an enantiomeric excess of at least 96% up to 100%, more particularly, having an enantiomeric excess of 98% up to 100%.

- 5 The stereoisomeric forms of the compounds of formula (I) have basic properties. The pharmaceutically acceptable acid addition salts as mentioned hereinabove comprise the therapeutically active non-toxic acid addition salt forms which the compounds of formula (I) are able to form. Said salt forms can conveniently be obtained by treating the base form of the compounds of formula (I) with appropriate acids such as inorganic
- 10 acids, for example, hydrohalic acid, e.g. hydrochloric, hydrobromic and the like acids, sulfuric acid, nitric acid, phosphoric acid and the like; or organic acids, such as, for example, acetic, propanoic, hydroxyacetic, 2-hydroxypropanoic, 2-oxopropanoic, ethanedioic, propanedioic, butanedioic, (Z)-2-butenedioic, (E)-2-butenedioic, 2-hydroxybutanedioic, 2,3-dihydroxybutanedioic, 2-hydroxy-1,2,3-propanetri-
- 15 carboxylic, methanesulfonic, ethanesulfonic, benzenesulfonic, 4-methylbenzenesulfonic, cyclohexanesulfamic, 2-hydroxybenzoic, 4-amino-2-hydroxybenzoic and the like acids. Conversely the salt form can be converted by treatment with alkali into the free base form.
- The term acid addition salt also comprises the hydrates and solvent addition forms
- 20 which the compounds of formula (I) are able to form. Examples of such forms are e.g. hydrates, alcoholates and the like.

- The four individual stereoisomeric forms of the compounds of formula (I) can be prepared by O-alkylating an enantiomerically pure phenol of formula (-)-(R)-(II) or
- 25 (+)-(S)-(II) with an enantiomerically pure 1,3-dioxolane derivative of formula (-)-(2S,cis)-(III) or (+)-(2R,cis)-(III) wherein -OR represents a sulfonyloxy leaving group such as 4-methylbenzenesulfonyloxy (tosylate) or methanesulfonyloxy (mesylate).

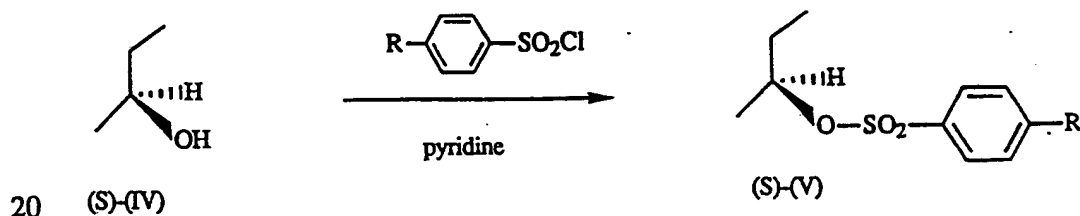


Said O-alkylation reaction can conveniently be conducted following art-known procedures, e.g. by stirring and heating the reactants in an appropriate solvent such as a dipolar aprotic solvent, e.g. N,N-dimethylformamide, N,N-dimethylacetamide and the like, in the presence of a base such as, an alkali metal hydroxide, e.g. sodium or potassium hydroxide. The thus obtained stereoisomeric forms of the compound of formula cis-(I) may be purified further following art-known methodologies such as liquid chromatography and crystallization.

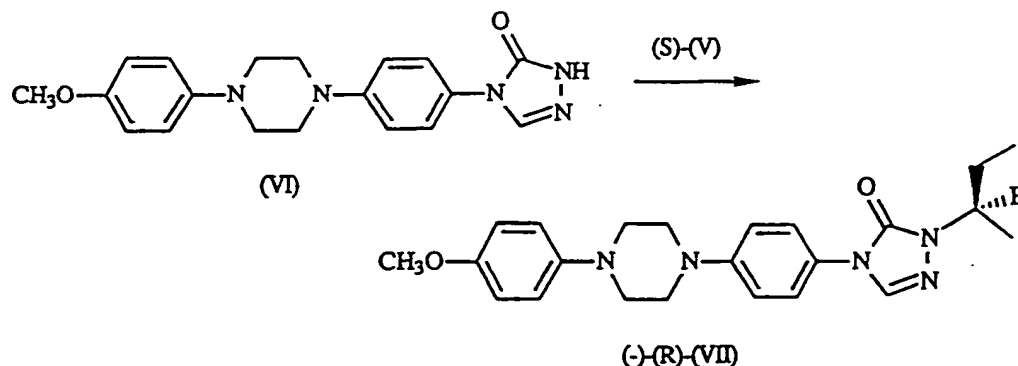
- 10 The relation between the stereochemistry of the stereoisomeric forms of the compound of formula (I) obtained in said O-alkylation reaction and the stereochemistry of the starting materials (II) and (III) is shown in the table herebelow.

cis-(I)	cis-(III)	(II)
(+)-[2R-[2 α ,4 α ,4(R)]]	(+)-(2R,cis)-(III)	(-)-(R)-(II)
(+)-[2R-[2 α ,4 α ,4(S)]]	(+)-(2R,cis)-(III)	(+)-(S)-(II)
(-)-[2S-[2 α ,4 α ,4(R)]]	(-)-(2S,cis)-(III)	(-)-(R)-(II)
(-)-[2S-[2 α ,4 α ,4(S)]]	(-)-(2S,cis)-(III)	(+)-(S)-(II)

- 15 The enantiomerically pure phenol of formula (-)-(R)-(II) can conveniently be prepared starting from (S)-2-butanol (IV). The enantiomeric (S)-butanol (IV) can be converted into a corresponding (S)-sulfonate (V) by reaction with 4-methylbenzenesulfonyl chloride (R = Me) or 1-bromo-4-benzenesulfonyl chloride (R = Br) in pyridine.

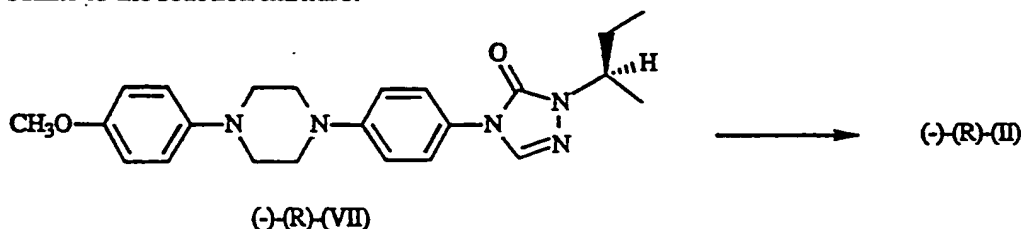


Enantioselective coupling of the sulfonate (S)-(V) with the triazolone (VI) (prepared as described in Example XVII of US-4,267,179) proceeds with inversion of configuration at the chiral center and yields (-)-(R)-(VII).



Said coupling reaction can be conducted in a reaction inert solvent such as a dipolar aprotic solvent, preferably *N,N*-dimethylformamide, in the presence of a base such as sodium hydride. The enantiomeric excess of the thus obtained product (-)-(R)-(VII) ranges from about 65% to about 75% and can be increased up to an enantiomeric excess e.e. > 98% by converting (-)-(R)-(VII) into the (R)-camphor sulfonate salt in acetone, followed by repeated recrystallization of the salt from ethanol / acetone mixtures (2:7; v/v).

Dealkylation of the thus purified anisole (-)-(R)-(VII) by heating to reflux in concentrated hydrobromic acid yields the enantiomerically pure phenol (-)-(R)-(II). In order to avoid bromination in the last mentioned step, it is advantageous to add sodium sulfite to the reaction mixture.



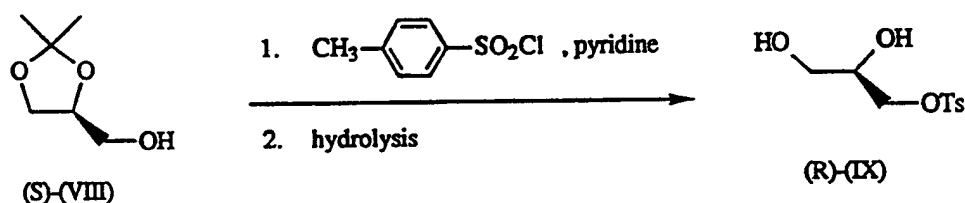
The other enantiomeric form, (+)-(S)-(II), can be prepared in a similar manner starting from (R)-2-butanol. Obviously, the optical purification of the thus obtained intermediate product (+)-(S)-(VII) is conducted on the (S)-camphor sulfonate salt.

An alternative procedure for preparing (-)-(R)-(VII) and (+)-(S)-(VII) comprises resolving the corresponding racemate (\pm)-(VII) by fractional crystallization with enantiomerically pure camphor sulfonic acid from a mixture of ethanol / acetone (1:4; v/v). Further optical purification to e.e. > 98% can be achieved by repeatedly recrystallizing the camphor sulfonate salt as described hereinbefore. (-)-(R)-(VII) is obtained from recrystallizing the racemate (\pm)-(VII) with (R)-camphor sulfonic acid; (+)-(S)-(VII) can be obtained similarly with (S)-camphor sulfonic acid from racemic (\pm)-(VII), or preferably from the mother liquors of the previous resolution step with (R)-camphor sulfonic acid which are enriched in the (+)-(S)-(VII) enantiomer.

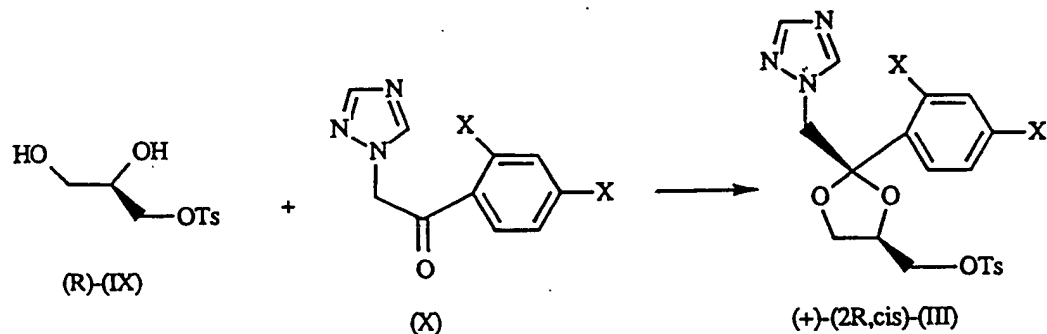
Preferably, however, the racemate (\pm)-(VII) can be resolved by liquid chromatography using a chiral stationary phase such as an amylose derivative, in particular amylose tris-(3,5-dimethylphenyl) carbamate coated on a macroporous γ -aminopropyl silica matrix (Chiralpak ADTM, Daicel), or a Pirkle type stationary phase. The separation is preferably conducted with an alcohol such as methanol or ethanol (optionally denatured with 1% methanol) as the mobile phase. In order to speed up the separation, the elution chromatography process may be conducted at a temperature higher than ambient (about 30°C). More time can yet be gained by using recycling peak shaving technology. For larger scale applications, the full continuous simulated moving bed adsorption technology can advantageously be used.

In an identical manner as described in the preceding paragraph, the racemate (\pm)-(II) can be separated into its two enantiomers on amylose tris-(3,5-dimethylphenyl) carbamate coated on a macroporous γ -aminopropyl silica matrix (Chiralpak ADTM, Daicel).

The enantiomerically pure intermediate (+)-(2R,cis) (III) wherein -OR is tosylate, can be prepared from the commercially available (S)-2,2-dimethyl-1,3-dioxolane-4-yl methanol (S)-(VIII). Reaction of (S)-(VIII) with 4-methylbenzenesulfonyl chloride in pyridine and hydrolysis of the thus obtained product in an aqueous acid solution, e.g. hydrochloric acid 6N, optionally in admixture with a solvent such as an alcohol or a ketone, e.g. acetone, yields the corresponding (R)-1,2-dihydroxypropyl tosylate (IX).



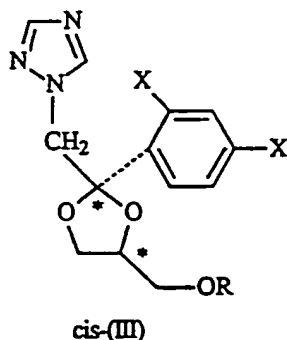
Acetalation of (R)-(IX) with 1-(2,4-dihalophenyl)-2-(1H-1,2,4-triazolyl)ethanone (X) under mild conditions yields a cis/trans mixture of dioxolanes from which the cis isomer (+)-(2R,cis)-(III) is obtained after chromatography.



Said acetalation reaction can conveniently be conducted by stirring and heating the reagents in a reaction-inert solvent, such as a halogenated hydrocarbon, e.g. dichloromethane or trichloromethane, in the presence of an acid, preferably a sulfonic acid, such as methanesulfonic acid.

In a similar manner, starting from (R)-2,2-dimethyl-1,3-dioxolane-4-yl methanol (R)-(VIII) there can be prepared (-)-(2S,cis)-(III).

The racemic compounds of formula (\pm)-cis(III)



wherein X represents chloro or fluoro and R represents 4-methylbenzenesulfonyl, methanesulfonyl, but also C₁₋₅alkylcarbonyl can conveniently be separated into their individual enantiomers by liquid chromatography using a chiral stationary phase such as a cellulose or amylose derived stationary phase. A preferred example of such a phase is cellulose tris-(4-methylbenzoate) either in its pure polymeric form or coated on a macroporous γ -aminopropyl silica matrix (Chiralcel ODTM, Daicel). The mobile phase preferably is an alcohol such as methanol or ethanol (optionally denatured with 1% methanol). Methanol is preferred with the cellulose derivative in the pure polymeric form. In order to speed up the separation, the elution chromatography process may be conducted at a temperature higher than ambient (about 30°C). More time can yet be gained by applying recycling peak shaving technology. For larger scale operations, the elution chromatography process is preferably replaced by full continuous simulated moving bed adsorption technology.

In case R represents C₁₋₅alkylcarbonyl, the separated enantiomers are converted into the mesylate or tosylate derivatives following art-known procedures such as saponification and subsequently sulfonylation.

The individual stereoisomeric forms of itraconazole and saperconazole have antifungal properties. Each of the four isomers contributes to the overall activity of the parent compound and none appears to be more active than said parent compound. The (-)-[2S-[2 α ,4 α ,4(R)]] and (-)-[2S-[2 α ,4 α ,4(S)]] isomers however appear more potent than their respective enantiomorphs.

The individual stereoisomers of itraconazole and saperconazole can also be obtained by separating the diastereomeric mixture by liquid chromatography on a chiral stationary phase such as an amylose derivative, in particular amylose tris-(3,5-dimethylphenyl) carbamate coated on a macroporous γ -aminopropyl silica matrix (Chiralpak ADTM, Daicel). Suitable mobile phases are alcohols in particular methanol and ethanol (optionally denatured with 1% methanol). In order to speed up the process and keep the product into solution, the separation is preferably performed at a temperature above ambient (up to 50°C). Limitations to the separation process are the fact that the compounds are poorly soluble and only small amounts can be processed at one time. Furthermore, the occurrence of 4 peaks precludes the use of recycling and peak-shaving technology. Consequently, separation of the intermediates for preparing stereochemically pure itraconazole and saperconazole is to be preferred over the separation of the end-products.

The individual stereoisomeric forms of itraconazole and saperconazole have greater water solubility than the diastereomeric mixtures of said compounds particularly when the individual stereoisomers are complexed with a cyclodextrin derivative.

Appropriate cyclodextrin derivatives are α -, β -, γ -cyclodextrins or ethers and mixed ethers thereof wherein one or more of the hydroxy groups of the anhydroglucose units of the cyclodextrin are substituted with C₁₋₆alkyl, particularly methyl, ethyl or isopropyl; hydroxyC₁₋₆alkyl, particularly hydroxyethyl, hydroxypropyl or hydroxybutyl; carboxyC₁₋₆alkyl, particularly carboxymethyl or carboxyethyl; C₁₋₆alkyl-carbonyl, particularly acetyl; C₁₋₆alkyloxycarbonylC₁₋₆alkyl or carboxyC₁₋₆alkyloxyC₁₋₆alkyl, particularly carboxymethoxypropyl or carboxyethoxypropyl; C₁₋₆alkylcarbonyloxyC₁₋₆alkyl, particularly 2-acetyloxypropyl. Especially noteworthy as complexants and/or solubilizers are β -CD, 2,6-dimethyl- β -CD, 2-hydroxyethyl- β -CD, 2-hydroxyethyl- γ -CD, 2-hydroxypropyl- γ -CD and (2-carboxymethoxy)propyl- β -CD, and in particular 2-hydroxypropyl- β -CD.

The term mixed ether denotes cyclodextrin derivatives wherein at least two cyclodextrin hydroxy groups are etherified with different groups such as, for example, hydroxypropyl and hydroxyethyl.

- 5 The average molar substitution (M.S.) is used as a measure of the average number of moles of alkoxy units per mole of anhydroglucose. In the cyclodextrin derivatives for use in the compositions according to the present invention the M.S. is in the range of 0.125 to 10, in particular of 0.3 to 3, or from 0.3 to 1.5. Preferably the M.S. ranges from about 0.3 to about 0.8, in particular from about 0.35 to about 0.5 and most
10 particularly is about 0.4.

- The average substitution degree (D.S.) refers to the average number of substituted hydroxyls per anhydroglucose unit. In the cyclodextrin derivatives for use in the compositions according to the present invention the D.S. is in the range of 0.125 to 3, in
15 particular of 0.2 to 2 or from 0.2 to 1.5. Preferably the D.S. ranges from about 0.2 to about 0.7, in particular from about 0.35 to about 0.5 and most particularly is about 0.4.

- More particular β - and γ -cyclodextrin hydroxyalkyl derivatives for use in the compositions according to the present invention are partially substituted cyclodextrin
20 derivatives wherein the average degree of alkylation at hydroxyl groups of different positions of the anhydroglucose units is about 0% to 20% for the 3 position, 2% to 70% for the 2 position and about 5% to 90% for the 6 position. Preferably the amount of unsubstituted β - or γ -cyclodextrin is less than 5% of the total cyclodextrin content and in particular is less than 1.5%. Another particularly interesting cyclodextrin derivative is
25 randomly methylated β -cyclodextrin.

- Most preferred cyclodextrin derivatives for use in the present invention are those partially substituted β -cyclodextrin ethers or mixed ethers having hydroxypropyl, hydroxyethyl and in particular 2-hydroxypropyl and/or 2-(1-hydroxypropyl)
30 substituents.

- The most preferred cyclodextrin derivative for use in the compositions of the present invention is hydroxypropyl- β -cyclodextrin having a M.S. in the range of from 0.35 to 0.50 and containing less than 1.5% unsubstituted β -cyclodextrin.
35

Substituted cyclodextrins can be prepared according to procedures described in U.S. Patent 3,459,731, EP-A-0,149,197, EP-A-0,197,571, US-4,535,152, WO-90/12035 and GB-2,189,245.

Other references describing cyclodextrins for use in the compositions according to the present invention, and which provide a guide for the preparation, purification and analysis of cyclodextrins include the following : "Cyclodextrin Technology" by József Szejtli, Kluwer Academic Publishers (1988) in the chapter Cyclodextrins in
5 Pharmaceuticals; "Cyclodextrin Chemistry" by M.L. Bender et al., Springer-Verlag, Berlin (1978); "Advances in Carbohydrate Chemistry", Vol. 12 Ed. by M.L. Wolfrom, Academic Press, New York (1978) in the chapter The Schardinger Dextrins by Dexter French at p. 189-260; "Cyclodextrins and their Inclusions Complexes" by J. Szejtli, Akademiai Kiado, Budapest, Hungary (1982); I. Tabushi in Acc. Chem. Research,
10 1982, 15, p. 66-72; W. Sanger, Angewandte Chemie, 92, p. 343-361 (1981); A. P. Croft and R. A. Bartsch in Tetrahedron, 39, p. 1417-1474 (1983); Irie et al. Pharmaceutical Research, 5, p. 713-716, (1988); Pitha et al. Int. J. Pharm. 29, 73, (1986); DE 3118218; DE 3,317,064; EP-A-94,157; US-4,659,696; and US-4,383,992. Particular attention should be paid to those references which describe
15 the preparation and purification methods which provide cyclodextrin mixtures wherein the amount of unreacted cyclodextrin is less than 5% of the total cyclodextrin content.

The complexes of the individual stereoisomeric compounds of formula (I) with cyclodextrin derivatives as mentioned hereinabove, can conveniently be prepared by
20 dissolving the cyclodextrin or ether derivative thereof in water and adding thereto the individual stereoisomeric compound of formula (I) while stirring or shaking the resulting mixture until complete dissolution is obtained. For storage purposes it may be advantageous to dehydrate the thus obtained solutions, e.g. by freeze-drying or spray-drying. The dehydrated complexes can easily be reconstituted by the addition of water
25 or an aqueous cyclodextrin solution.

The complexes of the individual stereoisomeric compounds of formula (I) with cyclodextrin derivatives are particularly useful for the preparation of pharmaceutical compositions for oral, parenteral or topical administration. Said pharmaceutical
30 compositions comprise as active ingredient a complex as defined hereinabove and a pharmaceutically acceptable carrier. Said carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirably in unitary dosage form suitable, preferably, for administration orally, rectally, percutaneously, or by parenteral injection. For example,
35 in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed, such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders,

disintegrating agents and the like in the case of powders, pills, capsules and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit form, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise
5 sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. In the compositions suitable
10 for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not cause a significant deleterious effect to the skin. Said additives may facilitate the administration to the skin and/or may be helpful for preparing the desired compositions. These compositions may be administered
15 in various ways, e.g., as a transdermal patch, as a spot-on, as an ointment. Acid addition salts of (I) due to their increased water solubility over the corresponding base form, are obviously more suitable in the preparation of aqueous compositions. It is especially advantageous to formulate the aforementioned pharmaceutical compositions in dosage unit form for ease of administration and uniformity of dosage. Dosage unit form
20 as used in the specification and claims herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such dosage unit forms are tablets (including scored or coated tablets), capsules, pills, powder packets, wafers, injectable solutions or
25 suspensions, teaspoonfuls, tablespoonfuls and the like, and segregated multiples thereof.

In the final compositions, the cyclodextrin will comprise about 2.5 to 20% by weight, in particular about 5 to 20%, more in particular 5 to 15%, for example about 10%, with the remainder being water, the active ingredient and any excipients.
30

In particular, stable pharmaceutical compositions may consist of water, cyclodextrin and the active ingredient alone without the need of additional stabilizers such as, human serum albumin, bovine serum albumin, lecithin, methyl cellulose, polyethylene glycol, sulfur containing reducing agents, urea, amino acids and surfactants. There may be
35 added a pH-adjusting agent e.g. hydrochloric acid, acetic acid, citric acid, sodium hydroxide, potassium hydroxide or a salt of any of these, in particular sodium citrate. The appropriate pH for formulating the itraconazole or saperconazole stereoisomers

ranges from 6.5 to 7.4, in particular from 6.8 to 7.0. Appropriate preservatives for the above pharmaceutical preparations are : alcohols, for example, ethanol, 1,3-propanediol, benzylalcohol or derivatives thereof, phenyl ethyl alcohol, phenol or phenol derivatives such as butylparaben, methylparaben, m-cresol or chlorocresol; acids, for example, 5 benzoic acid, sorbic acid, citric acid, sodium propionate, EDTA disodium; chlorhexidine; hexamidine diisetonate; hexetidine; optionally in combination with sodium bisulfite, or with propyleneglycol, or less preferably quaternary ammonium salts, metallic compounds such as zinc oxide, thiomersal and phenyl mercury salts, e.g. phenylmercuric acetate

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For the preparation of an injectable it is appropriate to add an isotonizing agent, e.g. sodium chloride, potassium chloride, sorbitol.

It may further be appropriate to add a suitable complexing agent such as calcium 15 chloride, citrate, EDTA and the like pharmaceutically acceptable metal ion complexing agents. For example, there may be added calcium chloride at a concentration of about 0.02 -2 g/l.

Said compositions can conveniently be prepared by dissolving the cyclodextrin or ether 20 derivative thereof in water and adding thereto the compound of formula (I) as well as other adjuvants and components such as, for example, sodium chloride, potassium nitrate, glucose, mannitol, sorbitol, xylitol and buffers such as, for example, phosphate, acetate or citrate buffer; and optionally concentrating or drying the solution by evaporation under reduced pressure or by lyophilization; and further optionally 25 reconstituting the lyophilized residue with water.

Preferred preparations according to the present invention have low toxicity, and are not irritating, thus permitting the manufacture of an injectable medicament which may safely 30 be used in repeated dose regimes without the risk of immunogenic reactions.

The aqueous preparations according to invention, and an excipient if required, may also be freeze-dried or spray-dried following art-known procedures to give a dehydrated composition which may be stored for a long period of time and dissolved before administration. In said freeze-dried or spray-dried formulations the molar ratio and the 35 weight-to-weight ratio of cyclodextrin to active ingredient may be the same as in the above-mentioned aqueous solutions. As it is convenient in a number of instances to reconstitute said freeze-dried or spray-dried formulation in an aqueous cyclodextrin solution, the molar ratio and the weight to weight ratio of cyclodextrin to active ingredient may also be lower than in the above mentioned aqueous solutions.

Experimental partA. Preparation of the intermediatesExample 1

- 5 To a solution of 40.5 g of 1-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-yl) ethanone in 300 ml of methanesulfonic acid and 300 ml of dichloromethane, were added dropwise 41.4 g of (S)-1, 2, 3-propanetriol 4-methylbenzenesulfonate (ester). The reaction mixture was stirred for 4 days at reflux temperature using a water separator. After cooling, the whole was added dropwise (over 1 hour) to a mixture of K₂CO₃ (3.6 M),
10 500 g of ice and 500 ml of trichloromethane while stirring. Stirring was continued for 1 hour and then the organic layer was separated. The aqueous layer was extracted with trichloromethane (3x) and the combined extracts were washed with water (2x) and evaporated. The residue was purified by column chromatography (silica gel ; CHCl₃). The eluent of the desired fraction was evaporated and the residue was converted into the
15 4-methylbenzenesulfonate salt (1:1) in 4-methyl-2-pentanone. The salt was filtered off and dried, yielding 27.0 g (26.9%) of (-)-(2S,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate (ester) 4-methylbenzenesulfonate (1:1 salt); mp. 194.5°C; $[\alpha]_D^{20} = -16.37^\circ$ (c = 1% in methanol) (interm. 1).

20

Example 2

- To a stirred solution of 12.8 g of 1-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-yl)-ethanone in 72.2 ml of dichloromethane were added dropwise 15 ml of
methanesulfonic acid during 15 minutes at room temperature. Upon complete addition,
25 14.8 g of (2R)-1, 2, 3-propanetriol 4-methylbenzenesulfonate (ester) were added and the whole was stirred and refluxed for 48 hours with the azeotropical removal of water. The previous mixture was added dropwise to a mixture of 195.5 ml of dichloromethane and a solution of 50 g of potassium carbonate in 200 ml of water. After stirring for a while, the separated organic layer was dried, filtered and evaporated. The residue was
30 purified by column chromatography over silica gel using trichloromethane as eluent. The pure fractions were collected and the eluent was evaporated. The residue was converted into the 4-methylbenzenesulfonate salt in 4-methyl-2-pentanone. The product was filtered off and dried, yielding 6.1 g (18.5%) of (+)-(2R,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methyl-
35 benzenesulfonate(ester) 4-methylbenzenesulfonate (1:1 salt); mp. 194.0°C; $[\alpha]_D^{20} = +16.44^\circ$ (conc. = 1% in ethanol) (interm. 2).

Example 3

- a) A mixture of 8.2 g of (R)-(-)-2-butanol, 31.5 g of 1-bromo-4-benzenesulfonyl chloride, 1.0 g of *N,N*-dimethyl-4-pyridinamine, 55.5 ml of pyridine and 293 ml of dichloromethane was stirred for 3 days at room temperature. After the addition of water, stirring was continued for 1 hour. The reaction mixture was washed twice with a diluted hydrochloric acid solution and once with water, dried, filtered and evaporated in vacuo. The residue was purified by column chromatography over silica gel using trichloromethane as eluent. The pure fractions were collected and the eluent was evaporated, yielding 11.5 g (35.6%) of (R)-2-butanol 4-bromobenzenesulfonate (ester) (interm. 3).

In a similar manner, (S)-(+)-2-butanol and 4-methylbenzenesulfonyl chloride were converted into (S)-2-butanol 4-methylbenzenesulfonate (ester) (interm. 4).

- b) A mixture of 3.5 g of 2,4-dihydro-4-[4-(4-methoxyphenyl)-1-piperazinyl]-phenyl]-3H-1,2,4-triazol-3-one, 0.6 g of a sodium hydride dispersion 50% and 100 ml of *N,N*-dimethylformamide was stirred for 3 hours at 80°C. After the addition of 3.5 g of (R)-2-butanol 4-bromobenzenesulfonate (ester) (intermediate (3)), stirring was continued for 6 hours at this temperature. After cooling, water was added. The crystallized product was filtered off and taken up in trichloromethane. The undissolved part was filtered off. The remaining solution was dried and purified by column chromatography over silica gel using trichloromethane as eluent. The pure fractions were collected and the eluent was evaporated. The residue was crystallized from 4-methyl-2-pentanone. The product was filtered off and converted into the (S)-7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-methanesulfonate salt in 2-propanone. The salt was filtered off and recrystallized from 1-butanol. The product was filtered off and dried, yielding 1.2 g (13.7%) of (+)-(S)-2,4-dihydro-4-[4-(4-methoxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (S)-7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-methanesulfonate (1:2); mp. 192.0°C; $[\alpha]_D^{20} = +28.41^\circ$ (conc. = 1% in ethanol) (interm. 5).

- In a similar manner, intermediate 4 was converted into (-)-(R)-2,4-dihydro-4-[4-(4-methoxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (R)-7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-methanesulfonate (1:2); mp. 193.0°C, $[\alpha]_D^{20} = -28.02^\circ$ (conc. = 1% in ethanol) (interm. 6).

c) A mixture of 29.4 g of (+)-(S)-2,4-dihydro-4-[4-[4-(4-methoxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (S)-7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-methanesulfonate (1:2) (intermediate (5)), 2.0 g of sodium sulfite and 151 ml of a hydrobromic acid solution 48% in water was stirred for 5 hours at reflux temperature. The reaction mixture was cooled to room temperature and water was added. The whole was neutralized with potassium carbonate to pH 7 while stirring in a mixture of dichloromethane and 1-butanol (90:10 by volume). The separated organic layer was dried, filtered and evaporated in vacuo. The residue was triturated in methanol. The precipitated product was filtered off and purified by column chromatography over silica gel using a mixture of trichloromethane and methanol (98:2 by volume) as eluent. The pure fractions were collected and the eluent was evaporated. The residue was crystallized from 4-methyl-2-pentanone. The product was filtered off and dried, yielding 10.4 g (77.7%) of (+)-(S)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 180.6°C; $[\alpha]_D^{20} = +4.38^\circ$ (conc. = 1% in methanol) (interm. 7).

In a similar manner, intermediate 6 was converted into (-)-(R)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 180.4°C; $[\alpha]_D^{20} = -4.16^\circ$ (conc. = 1% in methanol) (interm. 8).

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Example 4

A mixture of 44.6 g of 1-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-yl)ethanone, 56.0 g of (2S)-1,2,3-propanetriol 4-methylbenzenesulfonate (ester), 200 ml of methane-sulfonic acid and 150 ml of dichloromethane was stirred at reflux temperature using a water separator. After cooling, the reaction mixture was added dropwise to a mixture of ice-water, K_2CO_3 (aq.) and dichloromethane. The organic layer was separated and the aqueous phase was re-extracted with dichloromethane. The combined dichloromethane layers were dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; $CHCl_3$). The eluent of the desired fraction was evaporated and the residue was converted into the 4-methylbenzenesulfonate salt in 4-methyl-2-pentanone. The salt was recrystallized from 4-methyl-2-pentanone, yielding 20.5 g (16.4%) of (-)-(2S,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate (salt 1:1); mp. 182.5°C; $[\alpha]_D^{20} = -13.79^\circ$ (c = 1% in CH_3OH) (interm. 9).

35

Example 5

A mixture of 40.0 g of 1-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-yl)ethanone, 56.0 g of (2R)-1,2,3-propanetriol 1-(4-methylbenzenesulfonate) (ester), 250 ml of methanesulfonic acid and 100 ml of dichloromethane was stirred for 24 hours at reflux temperature using a water separator. After cooling, the reaction mixture was added dropwise to a mixture of K₂CO₃, water, ice and dichloromethane. The dichloromethane layer was separated, washed with water, dried, filtered and evaporated. The residue was purified by column chromatography (silica gel ; CHCl₃). The eluent of the desired fraction was evaporated and the residue was converted into the 4-methylbenzenesulfonate salt in 4-methyl-2-pentanone. The salt was recrystallized from acetonitrile, yielding 23.1 g (20.6%) of (+)-(2R,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1); mp. 183.5°C; $[\alpha]_D^{20} = +14.43$ (conc. = 1% in methanol) (interm. 10).

Example 6

A solution of 4 g of 2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one in 600 ml of methanol was resolved by liquid chromatography. Details of the chromatography process were :
column: 400 mm x 100 mm I.D. filled with about 1 Kg. 20 µm amylose tris-(3,5-dimethylphenyl) carbamate (Chiralpak ADTM: Daicel)
mobile phase : 150 ml/min. ethanol
temperature: 30 °C
The 4 g sample was separated in three cycles and yielded two fractions comprising 95-97% of the theoretical yield (mp. 158-162°C). Optical purity as determined by High Performance Liquid Chromatography was 97 % (interm. 11 and 12).

Example 7

A solution of 5 g (±)-cis-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate (ester) in 500 ml of methanol was resolved by liquid chromatography. Details of the chromatography process were :
column: 500 mm x 100 mm I.D. filled with about 2 Kg. 20 µm cellulose tris-(4-methylbenzoate) (Chiralcel OJTM: Daicel)
mobile phase : 150 ml/min. ethanol
temperature: 30 °C
The 5 g sample was separated in three cycles and yielded (-) compound (93-95% yield, HPLC purity : 100% ; α (365 nm) = -34.69° (1% MeOH) ; mp. 92-94°C) (interm. 13)

and (+) compound (95-96% yield, HPLC purity : 98.8% ; α (365 nm) = +34.02° (1% MeOH) ; mp. 92-94°C) (interm. 14).

B. Preparation of the final compounds

5 Example 8

A mixture of 5.2 g of (+)-(2R,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (2)), 2.9 g of (+)-(S)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (7)), 1.0 g of sodium hydroxide flakes and 100 ml of N,N-dimethylformamide was stirred for 7 hours at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off and taken up in trichloromethane. The organic layer was washed with water, dried, filtered and evaporated. The residue was purified by column chromatography (silica gel ; CHCl₃ / CH₃OH 98.5:1.5). The eluent of the desired fraction was evaporated and the residue was triturated in methanol, yielding 3.9 g (79.0%) of (+)-[2R-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 175.3°C; $[\alpha]_D^{20}$ = +19.08° (c = 1% in methanol) (comp. 1).

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Example 9

A mixture of 6.5 g of (-)-(2S,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (1)), 3.6 g of (+)-(S)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (7)), 1.0 g of sodium hydroxide flakes and 100 ml of N,N-dimethylformamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off, washed with water and taken up in dichloromethane. The organic layer was washed with water, dried, filtered and evaporated. The residue was triturated in methanol, yielding 4.6 g (72.5%) of (-)-[2S-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 146.5°C; $[\alpha]_D^{20}$ = -13.46° (c = 1% in methanol) (comp. 2).

35

Example 10

A mixture of 6.5 g of (-)-(2S,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (1)), 3.6 g of (-)-(R)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (8)), 1.0 g of sodium hydroxide flakes and 100 ml of N,N-dimethylformamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off, washed with water and taken up in dichloromethane. The organic layer was washed with water, dried, filtered and evaporated. The residue was triturated in methanol, yielding 4.4 g (69.3%) of (-)-[2S-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 156.6°C; $[\alpha]_D^{20} = -18.78^\circ$ (c = 1% in methanol) (comp. 3).

Example 11

A mixture of 6.5 g of (+)-(2R,cis)-2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (2)), 3.6 g of (-)-(R)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (8)), 1.0 g of sodium hydroxide flakes and 100 ml of N,N-dimethylformamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off, washed with water and taken up in trichloromethane. The organic layer was dried, filtered and evaporated. The residue was triturated in methanol. The product was dried in vacuo at 100°C, yielding 4.7 g (74%) of (+)-[2R-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 148.0°C; $[\alpha]_D^{20} = +14.15^\circ$ (c = 1% in methanol) (comp. 4)

Example 12

A mixture of 11.7 g of (-)-(2S,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (9)), 6.6 g of (+)-(S)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (7)), 1.5 g of sodium hydroxide flakes and 150 ml of N,N-dimethylformamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling,

water was added to the reaction mixture. The precipitate was filtered off, washed with water and taken up in dichloromethane. The organic layer was dried, filtered and evaporated. The residue was triturated in methanol, yielding 10.1 g (88.3%) of (-)-[2S-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 179.7°C; $[\alpha]_D^{20} = -8.41^\circ$ (c = 1% in methanol) (comp. 5).

Example 13

10 A mixture of 9.3 g of (-)-(2S,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate (ester) 4-methylbenzenesulfonate(salt 1:1) (intermediate (9)), 5.5 g of (-)-(R)-2,4-dihydro-4-[4-[4-[(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (8)), 1.0 g of sodium hydroxide flakes and 150 ml of N,N-dimethyl-
15 formamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, 800 ml of water were added. The precipitate was filtered off and taken up in dichloromethane. The organic layer was washed with water, dried, filtered and evaporated. The residue was purified by column chromatography (silica gel ; CHCl₃ / CH₃OH 98.5:1.5). The eluent of the desired fraction was evaporated and the residue was
20 triturated in methanol, yielding 7.7 g (81.8%) of (-)-[2S-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 183.3°C; $[\alpha]_D^{20} = -13.55^\circ$ (c = 1% in methanol) (comp. 6)

25 Example 14

A mixture of 9.9 g of (+)-(2R,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methylbenzenesulfonate (salt 1:1) (intermediate (10)), 5.5 g of (-)-(R)-2,4-dihydro-4-[4-[4-[(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (8)), 2.0 g of sodium hydroxide flakes and 150 ml of N,N-dimethyl-
30 formamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off and taken up in dichloromethane. The organic layer was dried, filtered and evaporated. The residue was crystallized from methanol, yielding 8.3 g (88.1%) of (+)-[2R-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]-methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 180.6°C; $[\alpha]_D^{20} = +9.00^\circ$ (c = 1% in methanol) (comp. 7)

Example 15

A mixture of 6.7 g of (+)-(2R,cis)-2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-yl-methyl)-1,3-dioxolane-4-methanol 4-methylbenzenesulfonate(ester) 4-methyl-
5 benzenesulfonate(salt 1:1) (intermediate (10)), 3.9 g of (+)-(S)-2,4-dihydro-4-[4-[4-(4-hydroxyphenyl)-1-piperazinyl]phenyl]-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one (intermediate (7)), 1.5 g of sodium hydroxide flakes and 100 ml of N,N-dimethyl-formamide was stirred overnight at 50°C under a nitrogen atmosphere. After cooling, water was added to the reaction mixture. The precipitate was filtered off and taken up in
10 dichloromethane. The organic layer was dried, filtered and evaporated. The residue was triturated with methanol, yielding 5.6 g (83.2%) of (+)-[2R-[2 α ,4 α ,4(S)]]-4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]-phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; mp. 182.4°C; $[\alpha]_D^{20} = +14.13^\circ$ (c = 1% in methanol) (comp. 8).

15

C. Physicochemical examplesExample 16

The solubility of the individual stereoisomeric forms of itraconazole and saperconazole in water, either in the absence or in the presence of different concentrations of
20 2-hydroxypropyl- β -cyclodextrin and at ambient temperature, was determined by preparing saturated solutions and measuring the amount of dissolved active ingredient. A 10 ml stock solution comprising 0, 2.5, 5, 7.5, 10, 20, 30 or 40% (weight by volume) 2-hydroxypropyl- β -cyclodextrin (M.S. = 0.4) was put in a 20 ml tinted vial. To each vial there was added a quantity of itraconazole or saperconazole sufficient to
25 yield a saturated solution. The vials were sonicated for 10 minutes and all vials were checked to ascertain that an undissolved remnant remained. Where all the material had dissolved, another quantity of the material was added. The vials were closed and protected from light by wrapping them in aluminum foil. The vials were then shaken in an appropriate apparatus. 24 Hours later the vials were checked again to ascertain that an
30 undissolved remnant of the material remained and where necessary a further quantity of the material was added. The vials were shaken for at least 72 hours and then allowed to stand until all undissolved material had precipitated. The vials were opened and the pH of the saturated solutions was measured. An aliquot of the supernatant solution was filtered and the amount of dissolved active ingredient was measured by UV
35 spectrometry (255 nm), if necessary after diluting said aliquot to a concentration suitable for UV spectrometry.

The tables hereinbelow summarize the concentrations (mg/100ml) of itraconazole and its individual stereoisomeric forms, and of saperconazole and its individual stereoisomeric forms, in water either in the absence or presence of different concentrations of 2-hydroxypropyl- β -cyclodextrin.

5

Table 1 (itraconazole)

active ingredient	stereochemistry	2-hydroxypropyl- β -cyclodextrin concentration							
		0%	2.5%	5%	7.5%	10%	20%	30%	40%
	cis (mixture of 4 isomers) itraconazole	n.d.	0.36	1.79	4.74	10.28	50	95	174
1	(+)-[2R-[2 α ,4 α ,4(S)]]	n.d.	2.77	16.21	39.64	78.98	354	889	1783
2	(-)-[2S-[2 α ,4 α ,4(S)]]	n.d.	3.58	16.36	41.39	78.16	332	846	1720
3	(-)-[2S-[2 α ,4 α ,4(R)]]	n.d.	2.01	14.21	37.64	96.40	371	1045	1860
4	(+)-[2R-[2 α ,4 α ,4(R)]]	n.d.	2.43	15.84	41.70	83.72	481	1166	2017

n.d. : not detectable by U.V. spectrometry

10

Table 2 (saperconazole)

active ingredient	stereochemistry	2-hydroxypropyl- β -cyclodextrin concentration				
		0%	2.5%	5%	7.5%	10%
	cis (mixture of 4 isomers) saperconazole	n.d.	0.97	4.62	8.92	16.92
5	(-)-[2S-[2 α ,4 α ,4(S)]]	n.d.	1.60	6.23	15.34	28.52
6	(-)-[2S-[2 α ,4 α ,4(R)]]	n.d.	1.13	5.05	13.42	26.51
7	(+)-[2R-[2 α ,4 α ,4(R)]]	n.d.	1.78	7.91	19.94	39.31
8	(+)-[2R-[2 α ,4 α ,4(S)]]	n.d.	2.02	8.15	18.14	36.31

n.d. : not detectable by U.V. spectrometry

15 Example 17

The solubility of compound 3 in artificial gastric juice was compared to that of itraconazole. About 10 mg of either product was added to 100 ml artificial gastric juice (0.2 g NaCl + 0.7 ml concentrated HCl diluted to 100 ml) at ambient temperature. The mixture was stirred and a sample was taken from each solution at regular intervals. The amount of dissolved active ingredient was determined following the procedure described

20

in the previous example. The dissolved amount of each ingredient (in mg / 100ml) is summarized in the table below.

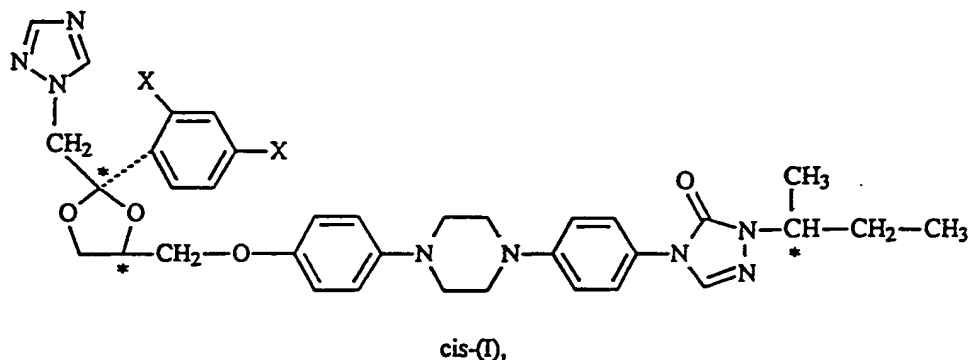
Time	concentration of dissolved itraconazole	concentration of dissolved compound 3
5 min.	n.d.	0.005
15 min.	n.d.	0.037
30 min.	n.d.	0.15
1 h.	0.023	0.32
2 h.	0.079	0.69
3 h.	0.15	1.09
4 h.	0.19	1.41
6 h.	0.27	1.94
26 h.	0.40	3.93
46 h.	0.40	5.08
51 h.	0.41	5.39
69 h.	0.45	6.36

n.d. : not detectable by U.V. spectrometry

Claims

1. A stereoisomeric form of itraconazole (X = Cl) or saperconazole (X = F) having the formula

5



or a pharmaceutically acceptable acid addition salt form thereof, wherein the (1H-1,2,4-triazol-1-ylmethyl) moiety and the substituted phenoxy moiety are located at the same side of the plane defined by the 1,3-dioxolane ring.

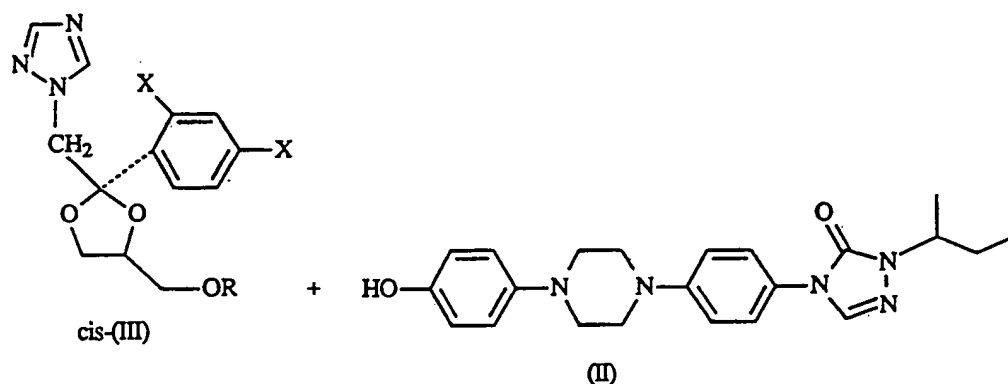
10

2. A stereoisomeric form according to claim 1 wherein said stereoisomeric form is
- (+)-[2R-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;
- (-)-[2S-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;
- (-)-[2S-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;
- (+)-[2R-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-dichlorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;
- (-)-[2S-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;
- (-)-[2S-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one;

(+)-[2R-[2 α ,4 α ,4(R)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one; or
 (+)-[2R-[2 α ,4 α ,4(S)]]-4-[4-[4-[4-[[2-(2,4-difluorophenyl)-2-(1H-1,2,4-triazol-1-ylmethyl)-1,3-dioxolan-4-yl]methoxy]phenyl]-1-piperazinyl]phenyl]-2,4-dihydro-2-(1-methylpropyl)-3H-1,2,4-triazol-3-one.

3. A stereoisomeric form according to claim 2 having a stereoisomeric purity of at least 96% up to a stereoisomeric purity of 100%.

4. A process of preparing a stereoisomeric form of itraconazole or saperconazole as defined in any one of claims 1, 2 or 3, characterized by Q-alkylating an enantiomerically pure phenol of formula (-)-(R)-(II) or (+)-(S)-(II) with an enantiomerically pure 1,3-dioxolane derivative of formula (-)-(2S,cis)-(III) or (+)-(2R,cis)-(III) wherein -OR represents a sulfonyloxy leaving group



thus yielding a stereoisomeric form of the compound of formula cis-(I), and optionally further purifying said compound and, if desired, converting said compound into a pharmaceutically acceptable acid addition salt form by treatment with an appropriate acid or conversely, converting the salt form into the free base form with alkali.

5. A complex comprising a stereoisomeric form of itraconazole or saperconazole as defined in claim 1 and an α , β , or γ -cyclodextrin or an ether or mixed ether derivative thereof.

6. A complex according to claim 5 wherein the cyclodextrin is a partially substituted β -cyclodextrin ether or mixed ether having hydroxypropyl or hydroxyethyl substituents.
- 5 7. A complex according to claim 6 wherein the cyclodextrin is hydroxypropyl- β -cyclodextrin having a M.S. in the range of from 0.35 to 0.50 and containing less than 1.5% unsubstituted β -cyclodextrin.
- 10 8. A process for preparing a complex as defined in any one of claims 5 to 7, characterized by dissolving the cyclodextrin or ether derivative thereof in water and adding thereto a stereoisomeric compound of formula (I), while stirring or shaking the resulting mixture until complete dissolution is obtained, and optionally dehydrating the thus obtained solution to a dehydrated complex.
- 15 9. A pharmaceutical composition comprising as active ingredient a therapeutically effective amount of a complex as claimed in any one of claims 5 to 7 and a pharmaceutically acceptable carrier.
- 20 10. A process of preparing a pharmaceutical composition as claimed in claim 9, characterized in that a therapeutically effective amount of a complex as claimed in any one of claims 5 to 7 is intimately mixed with a pharmaceutically acceptable carrier.
- 25 11. The use of individual stereoisomeric forms of itraconazole or saperconazole, or of their complexes with cyclodextrin or its derivatives, for preparing aqueous solutions wherein the concentrations of said ingredients are substantial greater than those attainable with the diastereomeric mixtures of itraconazole or of saperconazole.
- 30 12. A method of preparing aqueous solutions of itraconazole or saperconazole wherein the concentrations of said ingredients are substantially greater than those attainable with the diastereomeric mixtures of itraconazole or saperconazole, comprising the use of the individual stereoisomeric forms of itraconazole or saperconazole, or of their complexes with cyclodextrin or its derivatives.

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/EP 93/00552

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) ⁶		
According to International Patent Classification (IPC) or to both National Classification and IPC		
Int.Cl. 5 C07D405/14; A61K31/41		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁷		
Classification System	Classification Symbols	
Int.Cl. 5	C07D ; A61K	
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁸		
III. DOCUMENTS CONSIDERED TO BE RELEVANT⁹		
Category ¹⁰	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claims No. ¹³
A	US,A,4 916 134 (HEERES ET AL.) 10 April 1990 cited in the application see column 8, line 40 - line 51; claims ---	1-4,9
A	US,A,4 267 179 (HEERES ET AL.) 12 May 1981 cited in the application see column 22, line 47 - line 68; claims ---	1-4,9
A	US,A,4 764 604 (B.W.W. MÜLLER) 16 August 1988 cited in the application see claims -----	6-12
<p>¹⁰ Special categories of cited documents : ¹⁰</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.</p> <p>"A" document member of the same patent family</p>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search		Date of Mailing of this International Search Report
18 MAY 1993		28. 05. 93
International Searching Authority		Signature of Authorized Officer
EUROPEAN PATENT OFFICE		CHOULY J.

**ANNEX TO THE INTERNATIONAL SEARCH REPORT
ON INTERNATIONAL PATENT APPLICATION NO.**

EP 9300552
SA 70937

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report.
The members are as contained in the European Patent Office EDP file on
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